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Abstract

We have used a new, isentropic-coordinate three dimensional chemical transport model to investigate the decay of ClO and evolution of other species in the Antarctic polar vortex during September 1992. The model simulations coser to same southern hemisphere period studied in a companion data paper by Santee et al. [this issent. The model is initialized using the available data from the Microwave Limb Sounder (MES) and Cryogenic Limb Array Etalon Spectrometer (CLAES) on the Upper Atmosphere Research Smellite (UARS). During the model initialization chemical inconsistencies in the UARS data because evident. Fields of odd nitrogen (NO_y) derived from CLAES N_2O underestimated the sum of the direct observations of the major NO_y species. Results from the model integrations at 465 K and 585 K are sampled in the same way as the various UARS instruments and compared to the observations both directly and by considering average quantities in the inner and edge vertex regions. Sampling the observed species in the same way as the UARS instruments is important in removing any spurious trends due, for example, to changing solar zenith angle. While the model can reproduce the magnitude of the MLS CIO observations at 585 K, this is not possible at 465 K. The model partitions too much CIO into Cl_2O_2 to reproduce the observed CIO which is around 2.0 parts per billion by volume (ppbv) averaged within the polar vortex. The nodel also underestimates CLAES ClONO₂ in the inner vortex at 465 K due to heterogeneous processing. The observations require that effectively all of the inorganic chlorine is in the form of CiO) and ClONO₂ in the inner vortex at this altitude. In the basic model run, the decay of CIO produces CIONO₂ which is not observed by CLAES. Our results indicate the potential importance of the speculative reaction between OH and ClO producing HCl for the recovery of HC in the Autacetic spring. By including this

reaction, the decay of model ClO into HC 1 is cruanced, yielding better agreement with HCl data from the Halogen Occul tation Experiment (HA 1 OE) data Similar results can also be obtained by including the reaction between HO₂ and CHO o produce HCl with a 3% channel. The model generally reproduce τ_s the observed O_3 destruction during September. The most significant discrepancy for O_3 is all the inner vortex a 465 k where the model underestimates the observed O_3 loss—especially when the effects of vertical motion are included

observed decay of ClO was accompanied by an southern hemisphere. In the northern hemisphere the ern hemisphere and during September 1992 in the Sounder (MLS) during February 1993 in the next not matched by an increase in ClONO2. In fact the rine budget appeared closed. However, in the sout ern hemisphere the decay of ClO in September \mathbf{w} -Array Etalon Spectrometer (CLAES) and the dl crease in CIONO2 observed by the Cryogenic I vation of ClO as observed by the Microwave Liu (hereinafter referred to as \$96) discuss the dead of September 1992 using an off-line chemical tranat 585 K within the polar vortex. Because of this a CLAES CIONO2 data showed a slight negative tradi teorological analyses search Satellite (UARS) data and constrained by n port model initialized using Upper Atmosphere B parent discrepancy, in this paper we study the pensu In a companion paper, Santee et al. [this is u]

Off-line three-dimensional chemical transport notes els (CTMs) are becoming widely used in studys of stratospheric chemistry. When these models are forced by analyzed meteorological winds they provide a powerful framework for the interpretation of a wide range of stratospheric data [e.g., Hood of all 1989; Kaye et al., 1990; Lefèvre et al., 1994; Chappatheld et al., 1994]. In this study we use a new that dimensional CTM (SLIMCAT) which is a develotion mulated on hybrid sigma/pressure levels. The minimprovement of the SLIMCAT CTM over the TOM CAT CTM is the use of an isentropic vertical could nate with a radiation scheme to calculate the vertical motion.

It is important to understand quantitatively ClO deactivation in the polar regions as this will place a limit on the rapid chlorine-catalyzed O₃ depletion. In a recent paper, *Douglass et al.* [1995] commented on the different recovery (ClO deactivations regimes which occur in the northern and southern hemispheres. While the ClO decays into ClONO) in the northern hemisphere, this is not observed in the south. *Douglass et al.* [1995] argued that low O₃ causes the Cl concentration to increase relative to ClO permitting the faster recovery of HCl in the southern hemisphere; such possible conditions were also in the tioned by *Prather and Jaffe* [1990] and *Chipperfable* 1990].

The following section describes the SLLM CAT ($^{\circ}$ N

The model experiments are then described followed by details of the chemical initialization of the model from CARS data. The results of the model simulations are then presented followed by a summary.

Sld7 =: A' CTM

tails of the experiments discussed in this paper are the same vertical god as the prescribed wind and TOMCAT three-dimensional CTM first described by tropic three dimensional CTM is a development of the given at the end of this section. The SLIMCAT isena few general details are presented here. tical velocities are calculated by integrating the dibrid pressure/sigma levels as used in many general temperature fields which are either isobaric or hy-Chipperfield et al. [1993]. The TOMCAT CTM uses sis (or GCM) levels onto the model θ levels. temperature fields are interpolated from the analyspheric tracer transport. The horizontal wind and perature (θ) which are more appropriate for stratevertical levels are surfaces of constant potential temvergence of the horizontal winds. In SLIMCAT the circulation models (GCMs). of interest permitting high horizontal resolution. The can be run at just a few levels in a particular region very high vertical resolution. Alternatively, the model (or GCM) grid m cans that the model can be run with dence of the SLIMCAT vertical grid from the analysis the analysis (or GCM) vertical domain. The indepen $norm \theta$ levels which are completely contained within of SIAMCAT is limited by the maximum and mimthe analyses (or GCM) levels and the vertical domain SLIMCAT vertical grid is completely independent of number of layers, is a strength. The vertical, purely versatility of this model, the ability to use a small cal thermodynamic equilibrium (non-LTE) is not acalthough Curtis matrices are not used and so nonlomal infrared scheme is as described by Shine [1987] MIDRAD radiation scheme [Shine 1987]. The therdiabatic, motion in the model is calculated using the which produces unconnected isentropic levels also be run with the vertical transport switched off, account of the rest of the atmosphere. The model can levels above and below the SLIMCAT levels to take Rickaby [1989]. The radiation scheme uses dummy counted for. The solar scheme is that of Shine and As the CTM has not been described elsewhere, In TOMCAT the ver-

The model vertical grid is defined by specifying θ at the centers and interfaces of the model levels. The horizontal winds and temperatures (obtained from

will be implemented. a more physically based adjustment to the mass flux over most of the southern polar region. In the laters was performed using a tracer of mixing ratio 1 and experiments discussed below. the tracer was within 15% of the calculated beyoness not correcting the mass. of this correction has been examined for the 20 day is also adjusted in proportion to the total mass where keeping the mixing ratio constant. The magnitude ature field. During this readjustment the traces in se isentropic density obtained from the forcing tomper set the mass of the box to that calculated from the applied to each model box at every time step to reby MIDRAD. Therefore, an a posteriori confection is sarily be balanced by the vertical transport calculated gent/convergent horizontal mass fluxes will not be e-se mass at a point in the model caused by the diverwill not necessarily balance: the decrease/increase in zontal winds the transport of total mass in the model vertical winds is independent of the specified heri equal to that derived from the analysis temp to a box after a transport step will not necessarily mass as well as tracer mass and the mass content of The model advects (and conserves exactly) total air tor of 2 over the globe due to temperature variations the mass thickness of a level can change by a rect spectral transform. For a given isentropic hack at the horizontal box interfaces by choosing the con resolution and allows the mass fluxes to be calculated ficients allows a straightforward change of horizontal coefficients of the divergence and vorticity of the fluxes on isentropic levels. The use of spectral each mass fluxes which are then used to derive the spe is used to convert the horizontal winds to horizo tal contained in the model level. This mass thickness level interfaces and therefore the mass per unit sacra faces, permitting the calculation of pressure at the Temperature is also interpolated to the level in a els to the centers of the SLIMCAT isentropic l' 'l ses used here) are interpolated from the analysspectral coefficients in the case of ECMWI on Ly Furthermore, because the calculation of After 20 days the mass of A simple experiment

There are a number of benefits of using isenting levels for tracer transport in the stratosphere is quasi-horizontal motion in the stratosphere is a tropic, this coordinate system gives a true separate between horizontal and vertical motion. With a the between horizontal and vertical transport can be baric model the quasi-horizontal transport can be exchange of tracers between levels. As the vertical resolution of the model is likely to be fairly a resolution of the model is likely to be fairly a

lation procedure. which were continually being shocked by the assimicirculation, compared to the analyzed vertical winds scheme gave a better representation of the meridional lations of many months to years, use of a radiation products. Wearer et al. [1993] found that for simucussed the use of a radiation scheme to resolve some of calculation of heating rates. Weaver et al. [1993] disthe problems of vertical winds from assimilated data the vertical transport now depends on the accurate avoids this problem, although the representation of a radiation scheme to calculate the vertical transport the divergence) were noisy. In SLIMCAT the use of TOMCAT model, the vertical winds (derived from UARS pressure levels grid the interpolated porizontal winds u and v onto have a top level at 10 hPa). The UKMO analyses tracers are chemically active. Also, a large motiva-Mechum Range Weather Forecasts (ECMWF) which analyses available to us (e.g., European Centre for 1994]. The analyses extend much higher than other up to an altitude of 0.3 hPa [Swinbank and O'Neill, of UK Meteorological Office (UKMO) UARS analyses tion for the creation of SLIMCAT was the availability other species could be perturbed, especially if the not only will this lead to numerical diffusion but also When we used them in the

SLIMCAT uses the second-order moments advection scheme of Prother [1986], which is the scheme used in TOMCAT. This scheme is accurate and non-diffusive and well suited to the preservation of sharp gradients that can be produced in both the horizontal and the vertical, for example, during cases of polar stratespheric cloud (PSC) processing. The model can also be used with a first-order moments advection scheme of zero order moments scheme to requirement

SLIMCAT can be used simply as a transport model for possive tracers or with any idealized chemistry scheme. Alternatively, the model can be coupled with a detailed stratospheric chemistry scheme. The scheme used here is ideatical to that used in TOM-CAT and the species considered are listed in Table I. The chemical scheme has a treatment of heterogeneous chemistry on PSCs [Chipperfield et al., 1993, 1994] and sulfate acroscls [Chipperfield et al., 1995]. The occurrence of type 1 PSCs is predicted using the model fields of HNO₃. HyO, and temperature and the algorithm of Hanson and Manersberger [1988]. The model does not contain any microphysics or supersaturation: the PSCs occur whenever they are thermodynamically possible:

In this paper we have used the CTM to conduc experiments in the lower stratosphere for September 1992. As the region of interest is the lower strate sphere, we have used ECMWF analyses with the spetral truncation of T42 to force the model because these are available 6-hourly. In this region, diabates heating rates are small and so most experiments were performed without vertical transport. The model has izontal resolution was that of the Gaussian grid asser ciated with this spectral truncation (2.8°x2.8°). The model was run with three θ surfaces centered at 405 K, 585 K, and 655 K. The level at 655 K was in cluded so that descent into the layer at 585 K could be treated correctly. Photochemical data were gener ally taken from DeMore et al. [1994]. For the absorption cross sections of Cl₂O₂ the data of Burkholo v et al. [1990] was used in the wavelength region of 350-410 nm, which is the critical wavelength region for determining photolysis rates in the polar lover stratosphere.

Experiments

In the six experiments the model was integrated on three θ levels from 1200 UT August 31, 1992, for 20 days (32 days for runs B, D, and F). Run A was the basic model which included vertical motion. In run B, vertical motion was excluded; the three separate layers evolved independently. Experiment C was identical to experiment B except the photolysis of efficient for Cl_2O_2 was increased by a factor of 1.5 Experiment D was identical to experiment B except it included an 8% channel for reaction (R1) below. Finally, experiment E was identical to experiment B except it included a 0.3% channel for reaction (R3) below, while experiment F included a 3% channel for reaction (R3). The experiments are summarized if Table 2.

The net vertical motion produced in the model in run A was analyzed. On September 20, after 20 days integration, the net descent (change in θ) of air at 585 K was -20 K at the center of the vortex. At 465 K the maximum descent was -10 K at the edge of the vortex with less descent at center (not shown). These correspond to average vertical velocities of -0.45 mm/s at 585 K and -0.28 mm/s at 465 K. Because the vertical motion was small, run B was taken as the standard model to simplify the interpretation of results. Based on CLAES data, \$96 calculated the average vertical motion in the vortex during this period to be 0.5±0.3 mm/s at 585 K and -0.7±0.2 mm/s at 465 K and dis

cussed the effect of this on ClO_x (* $ClO + 2Cl_2O_2$) and $ClONO_2$.

Experiment C was performed to force the ClO:Cl₂O₂ partitioning in favor of ClO. As we have used the Cl₂O₂ cross section values of Burkholder et al. [1990] between 350 and 410 nm, which are larger than very recent recommendations [DeMore et al. 1994], experiment C is probably pushing J_{Cl2O2} above the limit of uncertainty. However, run C (discussed below), illustrates that even this increase in J_{Cl2O2} cannot sufficiently after the partitioning in favor of ClO to reproduce quantitatively the MLS ClO observations at 465 K.

Chandra et al. [1993] and Toumi and Bekki [1993] (using the balloon data of Stachnik et al. [1992]) have shown that inclusion of the following reaction

(R1) OH
$$\vdash$$
 ClO \rightarrow HCl \vdash O₂

as a minorchannel compared to

(R2) OH
$$\beta$$
 CIO (HO₂+C1

improves the agreement between 2D models and observations of ClO, HCl and O₃ depletion in the upper stratosphere (see also McElroy and Salawitch [1989] and Naturajan and Callis [1991]). Further, in an idealized study, Lary et at [1995b] suggested that this reaction could be important for the recovery of IICl after PSC processing. However, there is no direct chemical evidence for reaction (R1). It is the thermodynamically stable channel but involves a four-center transition state. Atkinson et al. [1995] suggested the branching ratio for this channel should be taken as $0.02_{-0.02}^{+0.12}$. In experiment D the effect of this reaction on the decay of lower stratospheric ClO was investigated quantitatively by comparison with UARS data. We adopted a yield of 8% for reaction (R1) similar to the previous modeling studies cited above.

Another speculative reaction which could lead to HCl formation is

(R3)
$$HO_2$$
 ClO -) $HCl + O_3$

as a minor channel compared to

(R4)
$$HO_2 \oplus ClO \longrightarrow O_2 \oplus HOCl$$

Although reaction (R3) is strongly exothermic it would also require a four-centered transition state. Burrows and Cox [1981] measured the upper limit for the quantum yield of this channel to be 0.3%, which is the current IUPAC recommendation [Atkinson et al., 1995].

Finkbeiner et al. [1995] have recently reported larger values which increase with decreasing temperature. At 210 K they measured the yield for reaction (R2) to be 5d 2%. The effect of this reaction was increasing temperature in experiments E and F. Experiment D was the currently recommended upper limit of 0.3% for the quantum yield (ff (R3), while run F used a value of 3%.

Initialization

For short case study integrations of CTMs the chemical initialization is critically important if the simulation is to be realistic. For this study the UARS data for August 31 was interpolated onto the D42 Gaussian grid on the isentropic surfaces θ . 465 K θ = 585 K, and θ = 655 K. The MLS O₃ data usec i. version 3. For H₂O, HNO₃ and ClO, data from more recent MLS retrievals are used, as described below The CLAES data is version 7 and the Halogen Oc cultation Experiment (HALOE) HCl data used be low for the model comparison is version 17 1Rus v li et al., 1996]. When HNO3 values depart substantially from climatology (e.g., following denitrification in the Antarctic polar vortex), MLS version 3 (20) values are too large by around 0.1 parts per billion by volume (ppbv) at 22 hPa and around 0.2 ppby at 46 hPa [Waters et al., 1996]. In this study we use ClO values from preliminary algorithms that also is trieve HNO₃, eliminating the bias in the version 3 data. Waters et al. [1995] also report a scaling coor in the ClO data due to the ClO line strength fact of being 8% greater than the value used in the carriest retrievals. We account for this by reducing the MIS ClO abundances by 8%. S96 give further details of the UARS data and details of the instrument errors. The UARS data are not entirely self-consistent and so the model initialization is not straightforward. All six experiments were initialized in the same way. In the absence of observations the CTM takes the initial chemical conditions from a 2D (latitude height) model initialized in potential vorticity (PV) - θ space using the method of Lary et al. [1995a] which was based on a technique first described by $School_{1}$ t tal. [1989].

The MLS and CLAES instruments on UARS make measurements on both the ascending and the discending side of the spacecraft orbit. Except during UARS yaw maneuvers, either the ascending or the descending side will correspond to mainly day time measurements and the other side will correspond

to mainly nighttime measurements. (During the yaw period both ascending and descending sides of the orbit can have significant numbers of daytime measurements). For the period studied here, the descending side corresponded to mainly daytime measurements up to September 3. After this date the ascending side made mainly daytime measurements. For shortlived species (e.g., ClO) it is important to distinguish between the ascending and the descending measurements on a given day, while for long-lived species (e.g., O₃) use of both ascending and descending sides together reduces the random errors in the data.

Long-Lived Tracers

UARS data for August 31 was used for the model initialization. CH₄ and N₂O were specified from CLAES data (averaged from the ascending and descending sides). O₃ was specified from (ascending and descending) MLS data [*Proidevaux et al.*, 1996]. B₂O was specified from the nonlinear MLS iterative retrieval [*Santee et al.*, 1995].

Odd Nitrogen Species

Observations from recent polar campaigns have shown the important role of measurements of long-lived tracers, such as N₂O, as a coordinate framework for other observations. Related to this are the correlations that are observed between pairs of long-lived tracers [Fahey et al., 1990; Plumb and Ko, 1992]. Measurements of one long-lived tracer can therefore be used to derive other long-lived tracers. In this way the CLAES N₂O could be used to derive the expected total odd netrogen (NO_r, in ppbv) from the expression

$$[NO_{c}^{*}] : 20.7 < 0.0644[N_{2}O]$$
 (1)

from Locwenstem et al. [1993] based on Arctic data. This expression is derived from ER-2 data and has not yet been validated at altitudes above 20 km. The above correlation will not apply if there has been denitrification/sedimentation which certainly occurs in the austral polar vortex during winter. This estimated NO_y can be compared with the sum of the individual MLS HNO₃ and CLAES NO₂ and ClONO₂. Figure 1a shows the derived NO_y* at 465 K. At high latitudes the maximum NO_y* is around 10 ppby. Between 60°S and 30°S there is a large region below 4 ppby and over the southern tip of South America NO_y* is slightly negative (-0.87 ppby, not shown by the chosen shading), which results from very high N₂O values. Around the 60°S latitude circle the sum of UARS

sion of 8%. By using the extreme of the CLAES can 15 ppby (Figure 2a). This is much larger than NO derived from CLAES N₂O in a region where denut species around $60^{\circ}S$ is in the wrong sense for this t ancy between NO_{y}^{*} and the sum of the UARS hitroge NO_y (and probably at 465 K too) and so the discret sum of UARS HNO₃ + NO₂ + ClONO₂ (Figure 2b) However, at 60°S the NO_y again underestimates the NO_y^* at high latitudes is around 15 ppbv (Figure 1b crepancy can be improved. At 585 K the maximu NO_y^* relationship [Locwenstein et al., 1993], this di limits with the extreme of the 15-20% error in the error in this CLAES N2O data is 20% with a pre- NO_y underestimates NO_y^* . The estimated systemat unlike at the center of the vortex when the measure! fication should not have invalidated the comparison $\mathrm{IINO_3} + \mathrm{NO_2} + \mathrm{ClONO_2}$ has a maximum of about At this altitude, denitrification would certainly reduc

Because of the apparent problems (due, no part, thigh N₂O values and the denitrification at the center of the vortex) the above correlation method vanot used. HNO₃ was initialized directly from (ascening and descending) MLS data [Santee et al., 1995] In the polar region the model NO_y was limited this MLS HNO₃ and the contribution from CIONO CLAES NO₂ was not used to initialize the model polyward of 45°S (see below).

Chlorine Species

The problems encountered above for the N₂O(NC)₂ correlation obviously implied that the CLALS N₂O could not be used to derive the expected inargamus chilorine (Cl_y) [e.g. Webster et al., 1993; Weadbridge et al., 1995]. Indeed, this procedure was tested and the derived Cl_y underestimated the sum of CLALS ClONO₂ and MLS ClO in the vortex region, especially at 465 K.

The model chlorine species were therefore initialized using a combination of the Cl_y field from the 2D model and the descending (daytime) measurements of MLS CIO and CLAES CIONO₂ [Mergenthales et al., 1996] and limiting the maximum total mixing ratio to 3.3 ppby. This sum was used as the total Cl_y unless the 2D model would have predicted more (for example, at the vortex edge). In the center and edge of the polar vortex, HCl was initialized to zero and any excess Cl_y after the initialization of first CIO and then CIONO₂ was added to CIO_x (CIO -1 $2Cl_2O_2$)

Aerosol Surface Area

The available across) surface area was specified in the model based on data from the SAGE II instrument (L. Thomason, personal communication, 1994). The initial surface area was constant with latitude and longitude and specified at 2 µm²/cm³ at 655 K, 5 µm²/cm³ at 585 K and 10 µm²/cm³ at 465 K. These values reflect the relatively high values that still persisted 15 months after the couption of Mount Pinatuho. As mentioned above, the occurrence of PSCs within each grid box is predicted on the basis of the model fields of H; O, HNO₃, and temperature.

homine Species

The bromine species were initialized from the 2D latitude height model. The upper stratospheric loading of BrO_y was 20 parts per trillion by volume (pptv) and at the center of the vortex at 465 K there was about 19 pptv. The short-lived bromine species rapidly adjust to the local photochemical conditions.

Lesu ts

area weighted mean of the measurements within these ner/edge vortex averaged values of MLS CIO and O3 585 K for September 5 and September 15. tour at 585 K. Figure 3 shows the calculated position the -25 PVU contour at 465 K and the -65 PVU condefined as the region between the inner vortex and 10^{-6} Km²kg⁻¹s⁻¹) contour at 465 K and the -130 the model. On each day the inner vortex region was defined by the limit of the -55 PVU (1 PVU = the UARS data by showing averages of the chemical species within the center and edge regions of regions and CLABS CIONO, were obtained by calculating an of the inner and edge vortex regions at 465 K and PVU contour at 585 K. The vortex edge was then from the ECMWF analyses that are used to force the vortex at 465 K and 585 K. We calculated PV We present comparisons between the model and

In the following plots the CIO and CIONO₂ averages were calculated from ascending data only while the O₃ averages were calculated using both ascending and descending data. For the HALOE HCl data the inner/edge vortex average is a simple mean of the individual survise profiles that fall within the defined PV contours on a given day. The model vortex averages were calculated in three ways. In the first method the average was an area-weighted value within the vortex at a fixed (1200 UT) time. Second,

for ClO and ClONO₂ the vortex average was also a culated by sampling the model at the same local time as the MLS and CLAES measurements throughout a 24-hour period. Third, for ClO a vortex average was also calculated from the model ClO sampled at 1.200 local solar time (LST) at all points.

Self-Consistency of UARS Initial Data

During the process of initializing the model, it became clear that there were chemical inconsistences between some of the UARS measurements. The problem with the CLAES N₂O has been mentioned above. Another example is MLS ClO and CLAES NO₂ at high southern latitudes on August 31 (not shown). There is a region of NO₂ greater than 2 pplex nor 330°E, 60°S and a large region exceeding 1 pplex a higher latitudes. These high values of NO₂ are concident with ClO values larger than 1 pplex. Following initialization with these two fields the model chemistry would lead to rapid production of ClONO₂ (and decay of ClO). Because the observed ClO does not decay so rapidly we did not use the CLAES NO₂ with in 10° latitude of the polar vortex.

Evolution of ClO, ClONO₂ and HCl

Vortex averaged quantities. Figure 4 shows the average model Cl_u species within the two vortex regions at 465 K and 585 K for run B (the standard model run without vertical transport). In this 100 the Cl_y is roughly constant at 2.8-3.0 ppbv at 465 K and 3.0-3.1 ppbv at 585 K. At both levels and in both in gions, ClO_x is decreasing during the model rate. The decay of ClO_x is more rapid in the edge vortex region compared to the inner vortex and more rapid at 585 K compared to 465 K. At 585 K the gradual decay of ClO_x in the inner vortex region produces an increase in ClONO₂ up to September 20, after which HCl recovers more rapidly. In the edge region at 5.5 K, ClO_x is lower than in the inner vortex and initially decays rapidly to ClONO2, although after September 6, ClONO₂ itself decays and HCl increases. At 465 K in the inner vortex the decay in model ClO, is initially slow. ClONO₂ increases slightly before September > 0 but after this date both ClONO₂ and HCl increase more rapidly. At the vortex edge at 465 K the decay of ClO₂ initially produces ClONO₂ with a small increase in HCl in late September.

Figure 5 shows the average MLS CIO inside the two vortex regions at 585 K and 465 K. (Note that the initial sharp increase in the MLS CIO is due to the changing LST of the measurements as the ascending

observations move from nighttime to daytime). The CIO data show larger average values in the inner vortex compared to the edge region and also larger values at 465 K relative to 585 K. At 585 K the inner vortex CIO shows a distinct decline after September 3, while in the edge region the much lower average values are more constant. At 465 K in the inner vortex the maximum average CIO values are near 2.0 ppbv and the inference of a trend is more difficult, although the values do decline after September 10. In the vortex edge at 465 K the average CIO values are relatively constant around 0.6 ppbv.

Also shown in Figure 5 are ClO results from model run B sampled at the same local time as the MLS measurements. The global model ClO fields were stored every 30 minutes and for each grid box the output closest to the MLS measurement time was used. By sampling the model at the same LST as the MLS measurements the vortex averaged ClO (after September 4) is significantly higher compared to the sampling at 12 UT (Figure 4) at both 465 K and 585 K as daytime CIO values are used at all longitudes. At 585 K this 'daytime' ClO is very similar in magnitude to ClO₂ (Figure 4) while at 465 K the 'daytime' ClO is significantly less than ClO_x (especially in the inner vortex) due to the contribution of Cl₂O₂. The decay of model ClO at 585 K in Figure 5 agrees very well with the MLS data in terms of both magnitude and trend in the inner vortex and slightly overestimates the magnitude in the vortex edge region. However, at 465 K the model underestimates the MLS observations by more than 0.5 ppbv in the inner vortex (Figure 5d). Interestingly, the decay of the model CIO at 465 K in Figure 5d (or indeed the CIO in Figure 4d) is not as strong as the decay in CIO₂. As CIO₂ decreases, the partitioning between ClO and Cl₂O₂ shifts in favor of ClO thereby reducing the apparent decrease

There may be also an influence on the observed CIO trend due to the changing LST of the MLS observations. The sharp increase in CIO before September 3 is due to the shift of the ascending observations from nighttime to daytime. S96 show that the solar zenith angic of the ascending CIO measurements then decreases from around 88° at high latitudes on September 3 to around 75° on September 17. Figures 5c and 5d also show the vortex averaged CIO from the model run B at 465 K when it is sampled at 12 LST at all longitudes. This sampling will maximize the model CIO (and minimize Cl₂O₂) for a certain amount of CIO₂ and illustrate the effect of the changing LST of

MLS observations on the apparent ClO trend '1 his sampling results in a significant increase in the model ClO in early September (when the ascending MLS observations were still mdarkness) but very similar values later in the month when the MLS measurements were essentially very close to local noor hither inner vortex the ClO sampled at 12 LS 1 shows very little change (although ClO₂ in Figure 43 degree at 12 LST dots \$ 10W amore distinct downward trend than ClO sampled like MLS, which implies that the changing MLS observation times disguises the (:0) trend slightly.

Figure 6 shows the average (; 10 DNO₂ from CLAES for the two vortex regions. At 585 K the CLAES data does not show a clear trend in the inner vortex w.) the average values remaining near 1.5 ppby In the edge region the CLAES observations appear to show a downward trend from around 1.7 ppby on August 3 to 1.4 ppby on September 17. At 465 K the (LAES vortex edge average remains around 1.1 pphy with me significant trend while inside the vortex the average ClONO₂ increases slightly from 0.5 to 0.7 ppby. In the single Vortex-fiver: Ir, c. analysis of \$96 the CLAES data showed a slight negative trend at 585 K and also at 465 K when the effects of vertical motion were accounted for.

Also shown in Figure 6 are model results from run B. At 585 K the model was sampled at both 12 UT and at the same LST'as the CLADS observations The difference between these 2 sampling patterns is around **0.1** ppby in the inner vortex and () 2 ppby in the edge vortex and illustrates the effect of the ClONO₂ diurnal cycle. At 465 K this effect is nea, ligible due to the slower photolysis at this altitude At 585 K the model (sampled like CLAES) produce... a significant mcrease in (,10)NO₂ in the innervoires. from 1.3 ppbv to 1.9 ppbv, which is not secure the CLAES observations. In the vortex edge—themonel dots show a decay from September 6 but themoel values are about 0.4 p pbv larger than CLADS At 465 K at the vortex $^{
m edge}$ the model C IONO, in 100 Bincreases from around 1.() ppby on September 1 to 1.5 ppby on September 17, an increase which ago in is not seem in the CL AES data. In the innervortx at 465 Krun B snows a slight increasing trotal 11 the model has much lower values than CLAES in this region which is still highly activated in the problem.

Generally, a number of features of run B show per comparison with the data. The model underestimate CIO at 465 K compared to MLS data and overest.

mates the recovery into ClONO₂ compared to CLAES at 585 K and in the vortex edge at 465 K. Conversely, the model underestimates ClONO₂ in the inner vortex at 465 K. These discrepancies were investigated in a range of sensitivity studies which are now discussed.

In run C, where the photolysis rate of Cl_2O_2 was increased by a factor of 1.5, the results at 585 K are very similar to run B and are not shown. Figures 5c and 5d (above) include the average ClO from run C at 465 K. Increasing J_{Cl2O2} has shifted the Cl_2O_2 :ClO equilibrium towards ClO and run C has around 0.2 pploy more ClO in the inner vortex compared to run B. The increase in ClO between run C and run B in the vortex edge (where ClO_x is lower) is much less. Despite this large 50% increase to J_{Cl2O2} the model still underestimates the MLS ClO by up to 0.5 pploy around September 10, which confirms the difficulties in quantitatively reproducing the observed values in the very low stratosphere.

By including the 8% channel for reaction (R1) in run D the results are significantly modified at both 465 K and 585 K. Figure 7 is a similar plot to Figure 4 but for run D. Although reaction (RI) is a sink of CIO the effect of this reaction depends on whether any further heterogeneous processing occurs. At 465 K compared to run B there is much more ClO_x , less CIONO₂, less HOCI and more HCl in run D. At this altitude there is ongoing PSC processing in the model runs. In run B the further ClO_x activation was limited by the low HCl abundance. In run D reaction (R)) forms HCt which can then be reprocessed by reaction with ClONO₂ to produce more ClO_x. At 585 K ClO_r in run D is similar to run B but there is less ClONO₂, more HCl and slightly less HOCl. At this altitude there is no further PSC processing and reaction (R1) favors the recovery of ClO₂ into HCl rather than ClONO2.

Results from run D are also compared with the MLS ClO and CLAES ClONO₂ data above (Figures 5 and 6). Compared to run B the reduced trend in ClONO₂ at 465 K at the vortex edge in run D is in better agreement with the CLAES data (see Figure 6 above), although the model average in the inner vortex at 465 K now underestimates CLAES by about 0.6 ppby. At 585 K results from run D (sampled at the same time as CLAES) are in better agreement with the observed magnitude and lack of trend in the inner vortex, although the model still shows a slight increase. In the vortex edge at 585 K run D also overestimates the CLAES data, but the discrepancy is less than run B. The evolution of ClO at 585 K in

run D is similar to run B both inside and at the by of the vortex and shows equally good agreement of the MLS CIO data (Figure 5). At 465 K, due to pongoing processing discussed above, run D has be haverage CIO than run B.

data also shows a real temporal change in HCI and the agreement with HALOE suggests that this estimate HALOB in the vortex center. The inere se HALOE data in the edge region but appears to over slightly after September 16 and then increases nor covery is initially slow. The rate of recovery increase in model HCl is obviously a true temporal increase for in the averaged quantities agrees well with the strongly after September 24. This temporal believe timates the HALOE data. At 465 K the model re both vortex regions and the model slightly underes from early September through to early October in recovery of the model vortex HCl in run D is steady gression of the HALOE observations. At 585 K the true temporal increase or caused by the polewant prodetermine whether this apparent increase in HCl is a crease in the inner vortex is similar. It is important to September 17 to 1.0 ppby on October 2 and the in strongly in the edge region from around 0.4 ppla on increases to 1.5 ppby. At 465 K the HCl increases K HCl is around 0.9 ppby on September 25 which September 17 to 1.5 ppby. In the inner vortex at 585 region the HCl increases from around 1.2 pply on is included in Figure 7. At 585 K in the vortex elp from September 17 onwards (see also S96 Figure 10) September 23. ber 17, and the inner vortex was first sampled on serve the southern vortex edge region until Sept in Due to its sampling pattern, HALOE did not of The vortex-averaged HALOL HO

Run E contained reaction (R3) as a 0.003% channel. The recovery of HCl (and evolution of other chlorine species) was only very slightly different to run B and the results are not shown here. Figure 8 shows the average Cl_y species from run F with the 3% channel for reaction (R3). In this run the effect of this 3% channel is similar to the effect of reaction (R1) in a m. D.

Synoptic maps. While the above analysis has been concerned with vortex-averaged quantities, so applie maps of chemical species from the model of also be compared with the observations. The model of discrepancies can then be investigated in terms of structure in the chemical fields in and around the polar vortex. As model run D gave the better agreement for the vortex-averaged quantities results from this run are used. However, for comparison with the

be reprocessed beterogeneously through reaction with the pole which is there not reprocessed due to the (HONO₂ in the collar region. absence of CIONO₂₇ near the vortex edge HCl can derestimates HALOE around 70°S. The higher model extends further south than the model plot). Although voltex average is caused by recovery of HCl nearer point-by-point comparison shows that the model unmates the simple HALOE HOlimer vortex average a the model vortex averaged HCl (Figure 7) overestivalues inside the vortex (note that the HALOE data relative minimum of HCl near 60°S followed by large at midlatitudes. Both run D and HALOE show a the mitial assumption of zero IICl in the vortex on mated in the model, which may be due, in part, to in IICl at the highest latitudes is slightly underestiservations (up to October 2). At 585 K the magnirun D sampled in the same way as the HALOE ob-August 31. At 465 K model run D overestimates HCl HALOE with values around 0.9 pphy. The increase tude of the model HCl at midlatitudes is similar to онеs October 5. Figure 9 also shows HCl from model sponding to September 12 and the highest latitude vations per day with the lowest latitude ones corre-October 1992. The map consists of around 15 obserand 585 K for the period in late September and early HALOE measurements of HCl interpolated to 465 K the duration of the experiment. Figure 9 shows the location and time as the HALOE observations over propriate and so the model was sampled at the same HALOD data a synoptic map on a given day is not ap-

decayed steadily but was still significant on Septemin the vortex in early September. The area of PSCs vortex. Inspection of the model results show that at the model with 0.25 - 0.75 pplor ClONO₂ in the inner observations again show a broader collar region than only slightly in the collar region and is still very low model has processed this air heterogeneously resulting 465 K the model predicted a large area of type I PSCs in the inner vortex. ber 15 (Figure 10d) the model CIONO₂ has increased vortex (and in some areas over 0.5 ppbv) while the sures over $0.25~\mathrm{ppbw}$ (3ONO₂ throughout the inner shows that at 465 K on September 5 CLARS meain very low mixing ratios in this region. By Septemvortex edge (Figures 6e and 6d above). in the inner vortex but similar in magnitude in the average CIONO₂ is significantly lower than CLAES synoptic maps of this species. At 465 K the model the model and CLALS is illustrated by considering The divergence between the average ClONO₂ from On September 15 the CLAES Figure 10b

ber 23. PSC activity finally ceased in late September, except for a small event ill early October. Figure 3 of S96 shows the acrosol extinction from CLAES '1 HI data shows PSCs were present in early September at 465 K but not letter in the month. The mode prediction of PSCs based on an equilibrium calculation (and an alyzed temperatures) appears to 01,,,,,,,,,, PSC activity at this altitude. The low ClONO2 the model inner vortex is a result of heterogeneous processing early in the run and the long times a conformal for ClONO2 recovery in the high latitude, denitrific i lower stratosphere.

At 585 K the vortex averaged ClONO₂ from tune! shows an increasing trend in the inner vortex, where the CLAES data shows no trend, and significant v overestimates the CLAES data at the vortex of a (Figures 6a and 6b above). These differences are elect in the synoptic maps. At 585 K on September 5 the model shows a continuous collar region with low values at the center (Figure 10a). The CLALS data shows higher values in the vortex with a limited collar region. By September 15 (Figure 10c) the data from CLAES still shows the same features as Septem ber 5 (the edge vortex average has decreased slightly) while in run D ClONO₂ has decreased in the color region but increased in the inner vortex. At 585 K the model only predicts type I PSCs from September 1 until September 3. On all these days the PSC vegion is small and located over the Palmer peninsura which is interesting in terms of the apparently anoma lous CLAES data in this region during this time.

Figure 11 shows a comparison between the III, be ClO fields from run 1) and the MLS data. The region of high ClO is generally well reproduced in the model at both altitudes. At 585 K the area of high ClO at the edge of the vortex is too large in the model in the magnitude of the highest ClO observed by M i Sour September 5 is well reproduced. The decayof (4C at 585 K in the model is slightly too slow. At 466 k Figures 11b and 11d again emphasize that the mode cannot reproduce the maximum observed values of ClO. In the run D the maximum model values are around 1.8 ppby inside the vortex, while 11, MLS values are over 2.25 ppby in certain regions of the vortex.

Ozone Loss

Figure 12 shows the average O₃ from experiments A, B, C and D along with the MLS data at the center and edge of the polar vortex at 465 K and 585 K. The MLS data shows the fastest rate of O₃ depletion

in the inner vortex at 465 K (around 60 ppbv/day), and a slightly slower rate in the edge region at this altitude. At 585 K the inner vortex ozone loss rate is significantly slower than the same region at 465 K. The MLS data shows that we are considering a period during which O₃ loss is ongoing but before very low (near zero) O3 mixing ratios occur. At the edge of the vortex all 4 experiments show similar loss rates which fit the data well at 465 K and slightly overestimate the loss at 585 K. In the inner vortex region model run B underestimates the depletion at 465 K and overestimates the depletion at 585 K. By including the effects of vertical motion (run A) the agreement is therefore better at 585 K but worse at 465 K following the descent of higher O₃ mixing ratio air. Compared to run B, run C gives a larger ozone depletion at 465 K (in better agreement with the observations) and very similar results at 585 K where the Cl₂O₂ cycle is less important for destroying ozone. Compared to run B, run D produces a stronger depletion at 465 K (due to higher (IO₂), again in better agreement with the observations, and slightly less depletion than run B at 585 K (due to faster deactivation). The underestimation of the observed O₃ loss in the inner vortex at 465 K is important as this is the region where there was the largest discrepancy between the model and observed ClO. If the model had higher ClO values in this region, similar to those measured by MLS, the modeled ozone loss would be larger.

Summary

We have used a new three-dimensional chemical transport model to study the evolution of chlorine species (and ozone) in the Antarctic polar vortex during September 1992. The model uses an isentropic vertical coordinate and the horizontal winds are specified from meteorological analyses or GCM output. Vertical advection is calculated using the MIDRAD stratospheric radiation scheme. The model uses the accurate, non-diffusive transport scheme of *Prather* [1986] and can be coupled with a detailed stratospheric chemistry scheme.

This modeling study complements a companion data paper by Santec et al. [this issue] which investigated the deactivation of chlorine in the two polar regions using UARS data. The model was initialized using observations from the UARS MLS and CLAES instruments and the evolution of the model was compared with the data.

There were obvious chemical inconsistencies be-

tween some of the UARS measurements. Fields of NO_y^* and Cl_y^* derived from CLAES N_2O underestimated the sum of individual members of these families. For this reason the powerful method of trace, correlation could not be used to derive these fields in the initialization procedure.

To compare the ClO fields between the model and measurements the model was sampled at the samblecal time as the MLS observations. This was also necessary for the comparison of ClONO₂ at 585 K because at this altitude ClONO₂ has a detectable disturbance. By sampling the model ClO like M15 certain observed trends in the ClO data can be as cribed to changing local solar times of the observations during early September.

A significant discrepancy between all of the model runs and the observations occurred for ClO at 465 K At 585 K the model could reproduce the magnitude of the MLS observations but at 465 K the model (40) strongly underestimated MLS CIO. The MLS CIO. cb. servations approach 2.4 ppbv in localized regions of the vortex and the inner vortex average at 465 K maximizes at 2.0 ppby. At 465 K the MLS data compare in magnitude to the model ClO₂ (ClO + 2Cl₂O₂) but the model partitions a considerable fraction of (10), into Cl₂O₂. A sensitivity run was performed (run \Box) in which the photolysis rate of Cl₂O₂ was increased by 50% to force the ClO:Cl₂O₂ partioning in favor of ClO at lower altitudes. However, this large increase in J_{Cl2O2} could not force the model to reproduce the large ClO mixing ratios observed by MLS

Another major discrepancy between the model and data occurred for ClONO₂. At 465 K, due to sugar ing heterogeneous processing and slow ClO recovery, there is very little ClONO₂ in the model inner vortex. In contrast the CLAES data indicates values around 0.5 pply. Together with the CIO comparis son mentioned above a major problem in the model is the simultaneous underestimation of both the CLAIS ClONO₂ and MLS ClO in the inner vortex at 465 kg. By simply adding these two observed quantities to g. Santee et al. [this issue], Figure 11) it can be seen that there is around 2.7 ppbv of chloring tied up in these two species alone. As the model only has 3.0 ppbv of Cl_v inside the vortex at 465 K it would be impossible to reproduce both the observed CIO and CIONO₂, after allowance is made for Cl₂O₂

The decay of CIO was studied in the basic model experiment and a number of sensitivity studies. In the basic model experiment (run B) CIO decays primarily into CIONO₂ at 465 K and in the inner vortex at 585

K which is not supported by the CLAES data. By including the reaction OH+ClO \rightarrow HCl+O₂ (RI) in run D the increase in ClONO₂ was suppressed and ClO₂ decayed into HCl. The recovery of HCl in run D was therefore in better agreement with the HALOE HCl data which indicated mixing ratios around 1.5 pply in the vortex in late September. We also studied the reaction HO₂+ClO \rightarrow HCl+O₃ (R3) in runs E and F. With a 0.3% channel the effect of this reaction in our experiments was negligible, but with a 3% channel the reaction increased recovery into HCl in a similar way to (R1).

The better agreement of run D with the HCl observations would support previous evidence [Toumi et al., 1993; Chandra et al., 1993] that reaction (R1) may occur with around an 8% channel. Uncertainties in the model initialization (especially of HCl) and UARS data preclude a more definite conclusion. Furthermore, the inclusion of speculative reaction (R3) with a 3% channel produced a similar effect on HCl recovery. Recent measurements [Finkbeiner et al., 1995] indicating a 5% yield for this channel would imply a very important role for reaction (R3) in the stratosphere.

Compared to MLS ozone the largest discrepancy in the depletion rate occurred in the inner vortex at 465 K. This was especially true when the effects of vertical motion were included. This underestimation of the O₃ depletion in this region is interesting in relation to the disagreements between the ClO and ClONO₂ fields discussed above. In the outer vortex at 465 K and at 585 K the model depletion rate was generally in good agreement with the MLS data.

The numerical model experiments presented here have aided the interpretation of the UARS data presented by Santee et al. [this issue]. Models are able to quantify the sampling effects of diurnally varying species and test for the chemical self-consistency of different datasets. The comparison between model and data show some quantitative discrepancies. To further investigate these will require more constraint by improved UARS retrievals leading to better model initialization.

Acknowledgments MPC thanksr. Simonof Météo France for Felp with the transport model, 1<1. Shine and J Thuburn for supplying the MIDR AD radiations cheme, J. A. Pyle for helpful comments and 1'. Berrisford for his assistance in obtaining the ECMWF analyses. We acknowledge use of the ECMWF analyses. The modeling work at Cambridge University

forms part of the UK Universities' Global Atreaspheric Modelling Programme and is also supported by the U.K. Department of the Environment

References

- Atkinson, R., D.L. Baulch, R.A. Cox, R.F. Hamps on J.A. Kerr, and J. Troe, Evaluated kinetic and picotochemical data for atmospheric chemistry, Supplement V IUPAC Subcommittee on Gascki netic Data Evaluation for Atmospheric Chemistry, J. Phys. Chem. Ref. Data, (in press), 1995.
- urkholder, J.B., J.J. Orlando, and C.J. Howard Ultraviolet absorption cross sections of Cl₂O₂ to tween 210 and 410nm, *J. Phys. Chem.*, 94, 687–695 1990.
- urrows, J.P., and R.A. Cox, Kinetics of chlorace axide radical reactions using modulated photolysis Part 4. The Reactions of Cl + Cl₂O -> Cl₂ + Cl₂O and ClO + HO₂ -> products studied at 1 at no and 300 K, J. Chem. Soc. Paraday Trans. 77, 24–5, 1981.
- Chandra, S., Jackman, C. H., Douglass, A. R., Elconing, E. L., Considine, D. B., Chlorine catalyzed destruction of ozone implications for ozone variability in the upper-stratosphere, *Geophys. Res. Lett.* 20, 351-354, 1993.
- Chipperfield, M.P. J. A. Pyle, C. E. Blom, N. Glatthor, M. Höpfner, T. Gulde, Ch. Piesch and P. Simon, The Variability of ClONO₂ in the Arctic Polar Vortex: Comparison of Transall MHTAS Measurements and 3D Model Results, *J. Geophys. Res.*, in press, 1995.
- Chipperfield, M.P., A numerical study of trace gases in the polar stratosphere, PhD Thesis, University of Cambridge, 1990.
- Chipperfield, M.P., D. Cariolle, P. Simon, R. Ramaro son and D.J. Lary, A three-dimensional model of study of trace species in the Arctic lower strate sphere during winter 1989-90, J. Geophys. Bis 98, 7199-7218, 1993.
- Chipperfield, M.P., D. Cariolle and P. Simon, 2, 31 chemical transport model study of chloring settler tion during EASOP, *Geophys. Res. Lett.*, 21, 1–7, 1470, 1994.

- DeMore, W.B., et al., Chemical kinetics and photochemical data for use in stratospheric modeling JPL Publication 94-26, 1994.
- Douglass, A.R., M.R. Schoeberl, R.S. Stolarski, J.W. Waters, J.M. Russell, A.E. Roche and S.T. Massie, Interhemispheric differences in springtime production of HCl and CiONO₂ in the polar vortices, J. Geophys. Res., 100–13,967-13,978, 1995.
- alicy, D.W., S. Solomon, S.R. Kawa, M. Loewenstein, J.R. Podolske, S.D. Strahan and K.R. Chan, A diagnostic for denitrification in the winter polar stratospheres. *Nature*, 345, 698-702, 1990.
- Finkbeiner, M., J.N. Crowley, O. Horie, R. Muller, G.K. Moortgat and P.J. Crutzen, Reaction between tween HO₂ and ClO: Product formation between 210 and 300 K, *J. Phys. Chem.*, 99, 16264-16275, 1995.
- Hoidevaux, L., Cal., Validation of UARS MLS O₃ Measurements. J. Geophys. Res., (in press), 1995.
- lanson, D., and K. Mauersberger, Laboratory studies of the nitric seil trihydrate: Implications for the south polar stratosphere, *Geophys. Res. Lett.*, 15, 855–858, 1988.
- Kaye, J.A., et al., Three dimensional simulation of hydrogen eldoride and hydrogen fluoride during the Airborne Arctic Stratospheric Expedition, Geophys. Ites. Lett., 17, 529-532, 1990.
- Norton and L.P. Riishojgaard, Three-dimensional tracer initialization and general diagnostics using equivalent PV latitude potential temperature coordinates, Q. J. R. Meteorol. Soc., 121, 187-210, 1995a.
- Lary, D.J., M.P. Chipperfield, and R. Toumi, The potential impact of the reaction OH4 ClO-5HCl4 O₂ on polar ozone photochemistry, *J. Atmos. Chem.*, 21, 61-79, 1995b.
- Lefevre, L., G.P. Brasseur, I. Folkins, A.K. Smith and P. Simon, Chemistry of the 1991-1992 stratospheric winter: Three dimensional model simulations, J. Geophys. Res., 99, 8483-8495, 1994.
- ocwenstein, M., et al., New observations of the NO_y/N₂O correlation in the lower stratosphere, Geophys. Res. Lett., 20, 2531-2534, 1993.

- Mergenthaler, J.L., et al., Validation of CLAFS ClONO₂ measurements, J. Geophys. Res., 4in press), 1995.
- Natarajan, M., and L.B. Callis, Stratospheric photochemical studies with Atmospheric Trace Molecule (ATMOS) measurements, J. Geophys. Res., 95 9361-9370, 1990.
- Plumb, R.A., and M.K.W. Ko, Interrelationships between mixing ratios of long-lived stratospheric constituents, J. Geophys. Res., 97, 10,145-10,156-1992.
- Prather, M.J., Numerical advection by conservat in of second order moments, J. Geophys. Res. 97-6671-6681, 1986.
- Prather, M.J., and A.H. Jaffe, Global impact of the Antarctic ozone hole: chemical propagation, J. Geophys. Res., 95, 3473-3492, 1990.
- Roche, A.E., et al., Validation of CH₄ and N₂O m a surements by the CLAES instrument on UARS J Geophys. Res., (in press) 1995.
- Rood, R.B., D.J. Allen, W.E. Baker, D.J. Lamich and J.A. Kaye, The use of assimilated strato-pheric data in constituent transport calculations, J. Atmos. Sci., 46, 687-701, 1989.
- Russell, J.M., et al., Validation of hydrogen chloraic measurements made by HALOE from the UARS platform, J. Geophys. Res., (in press) 1995.
- Santee, M.L., et al., Chlorine deactivation in the lower stratospheric polar regions during late winter: results from UARS, J. Geophys. Res., (submitted), 1996.
- Santee, M.L., W.G. Read, J.W. Waters, L. Proide-vaux, G.L. Manney, D.A. Flower, R.F. Jarnot, R.S. Harwood and G.E. Peckham, Interhemispheric differences in polar stratospheric HNO₃, H₂O, €:0 and O₃, Science, 267, 849-852, 1995.
- Schoeberl, M.R., et al., Reconstruction of the constituent distribution and trends in the Antarctic polar vortex from ER-2 flight observations. J. G. o. phys. Res., 94, 16,815-16,845, 1989.

- Shine, K.P., The middle atmosphere in the absence of dynamical heat fluxes, Q. J. Roy. Meteorol. Soc., 113, 603-633, 19-87
- Shine, K.P., and J.A. Rikaby, Solar radiative heating due to absorption by ozone, 597-600, Proceedings of Quadrennial Ozone Symposium, Got tingen, A. Deepak Publishing 1989.
- Stachnik, R. A., J. C. Hardy, J. A. Tarsala, J.W. Waters and N.R Erickson, Submillim eterwave heterodynemicas name is & stratospheric ClO, HCl, O₃ and HO₂; first results, *Geophys. Res. Lett.* 19, 1931-1934, 1992
- Swinbank, R. and A.O. Neill, A. stratosphere-troposphere data assimilation system, *Mon. Wea. Rev.*, 1: 2, 686-702, 1994.
- Tourni, R. and S. Bekli, The importance of the reactions between OH and ClO for stratospheric ozone, Geophys. Res. Lett., 10, 2447-2450,1993.
- Waters, J. W., et al Validation of UARS M1 s. ClO measurements, J. Chophys. Res., 100, (in press), 1995.
- Weave r, C.J., A.R 1 Douglass and R.B. Rood, Thermodynamic balance of three-dimensional stratospheric winds lerived from a data assimilation procedure, J 44/1/20, S. 11, 50, 2087-2993, 1993.
- Webster, C.R. et al. Chlorine chemistry on polar stratospheric cloud particles in the Arctic winter, Science 26 1, 1130 1134, 1993.
- Woodbridge, I.I., et al., Estimates of total organic and inorganic chlorine in the lower stratosphere from in situ and flask measurements during AASE 11, J. Grophys Res. 100, 3057-3064, 1995.
- Table 1. Chemical Spicies Contained in the Model

 $\begin{array}{l} {\rm Transported} \\ {\rm O}_{x} \ (= {\rm O}_{3} + {\rm O}(^{3}{\rm P}) + {\rm O}(^{2}{\rm I}))), \ {\rm H}_{2}{\rm O}_{2} \\ {\rm NO}_{z} \ (= {\rm NO}_{+} {\rm NO}_{2} + {\rm NO}_{3}), {\rm N}_{2}{\rm O}_{5}, {\rm H}{\rm NO}_{3}, {\rm HO}_{2}{\rm NO}_{2}, \\ {\rm ClO}_{z} \ (= {\rm Cl}_{+} \ (^{1}{\rm o} + ^{2}{\rm Cl}_{2}{\rm O}_{2}), {\rm ClONO}_{2}, {\rm HCl}, {\rm HOCl}, {\rm OClO}, \\ {\rm BrO}_{z} \ (= {\rm Br} + {\rm BrO}), {\rm BrONO}_{z}, {\rm BrCl}, {\rm HBr}, {\rm HOBr} \\ {\rm N}_{2}{\rm O}, \ {\rm ClI}_{4}, \ {\rm COH}_{2}{\rm O} \end{array}$

(

Table 2. Summary of SLIM CAT Model Experiments

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	Run	Vertical Transp	ort J _{Cl2O2} Factor	- Quantum Yield - (R1) (R3)	
	Λ	yes	J	0	0
]}	110	1	0	()
	('	во	$\pm .5$	0	0
Figure 1. Estimate of NO _y obtained from correlation with b) 585 K on August 31 1992.	$\operatorname{Cl}_E^{\operatorname{D}}\operatorname{E}$	No data using	equation (1) for a) 4	$65_0^{0.08}$ an	$d\frac{0}{0.003}$
	F.	no	1	0	0.03

Figure 2. Sum of MLS HNO₃ and CLAES CHONO₂ and NO₂ (ppbv) for a) 465 K and b) 585 K on August 31 1992.

Figure 3. Position of the polar vortex (based on PV calculations) showing the inner vortex (dark shading) and the vortex edge (lighter shading) for a) 585 K September 5 b) 465 K September 5 c) 585 K September 15 and d) 465 K September 15.

Figure 4. Vortex averages of model ClO, ClO_x, HOCl, ClONO₂, HCl and Cl_y (pplw) sampled at 1200 UT versus day (after August 31 1992) from run B for a) 585 K edge vortex, b) 585 K inner vortex, c) 465 K edge vortex and d) 465 K inner vortex.

Figure 5. Vortex averages of ascending MTSCIC+T>) and model (10 (pploy) tota) 585 K vortex edge, b) 585 I{ inner vortex, c) 465 k vortex edge and d) .65 K it mer vortex. The mode t has been sampled at the same local time as the MLS measurements for cachgrid, nont total 465 K run B has also been sampled at 1200 local time at all longitudes. In addition results from run Calc shown at 465 K at the results from run D at 465 K and 585 K (sampled like MLS).

Figure 6. Vortex averages of ascending CLAES GUNO2 (asterisks) and model CiONO2 (ppbv) from run B and run D for a) 585 K vortex edge, b) 585 K inner vortex. [465 K vortex edge, and d) 465 K inner vortex. At 585 K the model CiONO2 has been sampled at both 1200 U = and at the local time of the CLAES measurements. At 465 K the diurnal variation of CiONO2 is negligible and the model is only sampled at 1200 UT. Note there is no CLAES data for September 1.

Figure 7. As figure 4 but for run D. Also shown scalic vartex-averaged HCl from HALOE (4 marks).

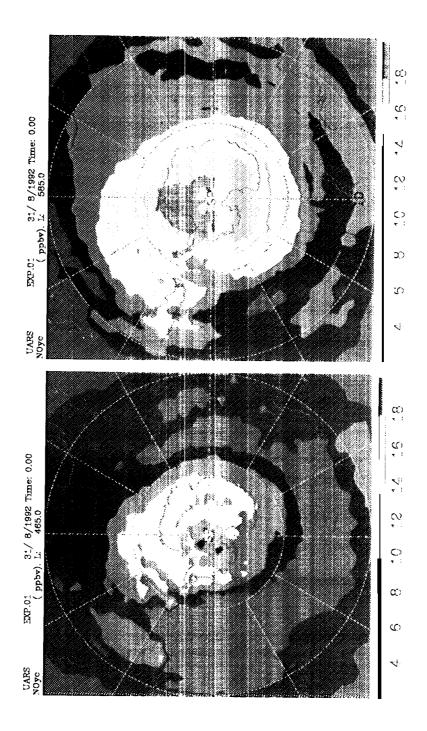
Figure 8. As figure 4but for run F.

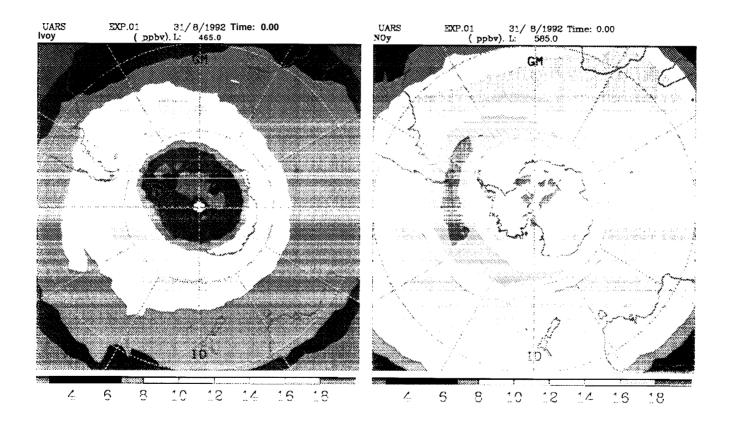
Figure 9. HALOE observations of HCl (pplw) interpolated to a) 585 K and b) 465 K. There are around 15 observations per day which progress southwards with the c; the low latitude observations were made on September 12 and the highest latitude ones on October 5. Results from model run D (sampled at the same time and location as the HALOE observations) from September 13 to October 2 are shown in figures 8c and 8d.

Figure 10. Distribution of CIONO₂ from CLAES and model run D (sampled at the same local time as CLAES) for a) 585 K September 5, b) 465 K September 5, c) 585 K September 15 and d) 465 K September 15. The model fields are discontinuous at 180°E due to the 24 horn shad in UT across the date line. Bad data points have been removed from the CLAES maps and show up as across of zeros, e.g. near the Palmer peninsula in figure 9a.

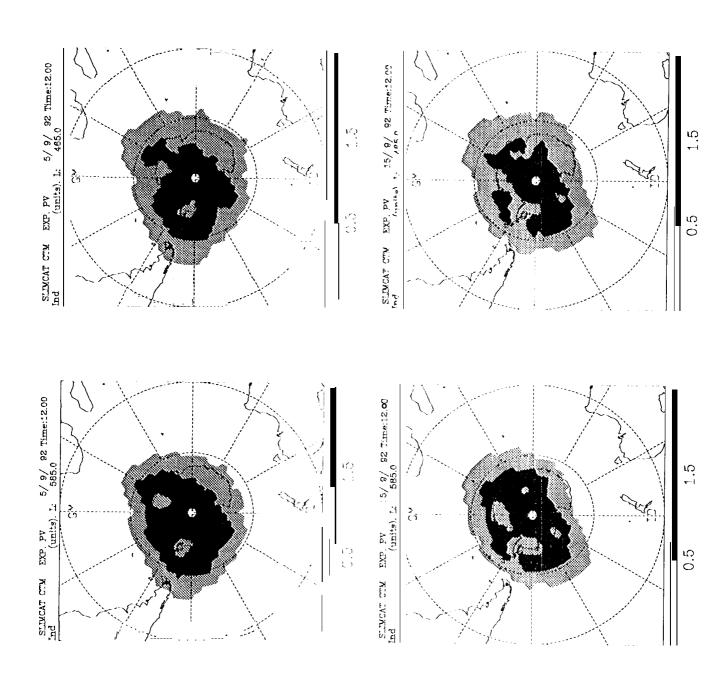
Figure 11. Distribution of ClO from MLS and model r in D (sampled at the same local time as MLS) for a) 585 K September 5, b) 465 K September 5, c) 585 K September 15 and d) 465 K September 15. The model fields are discontinuous at 180°E due to the 24 hour shift in UT a ross the date line.

Figure 12. Vortex averages of MLS O₃ (triangles) and model O₃ (ppinv) from runs A,B,C and D for a) 585 K vortex edge, b) 585 K inner vortex, c) 465 K vortex edge, and d) 465 K inner vortex.





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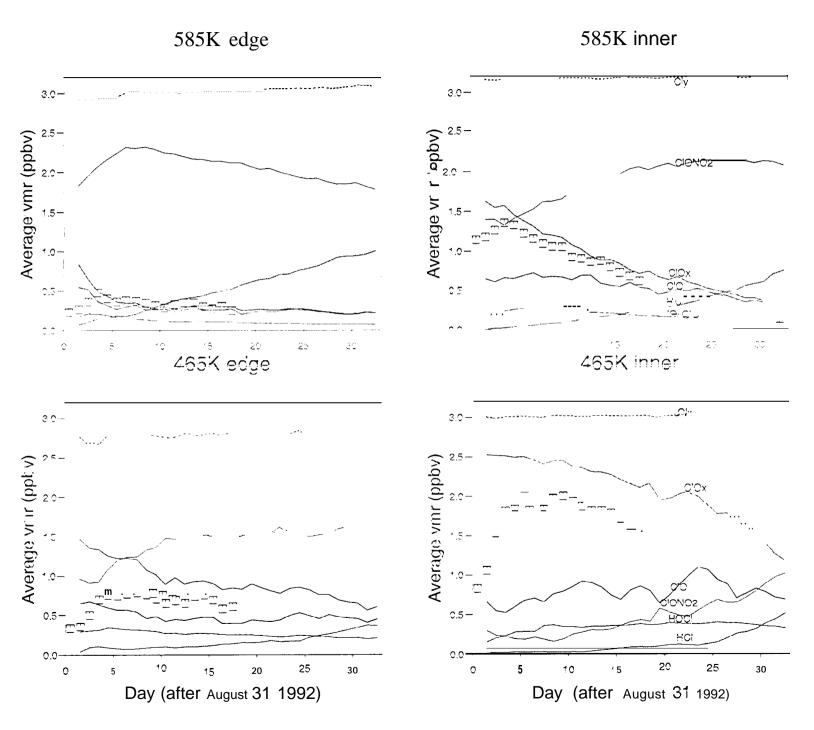
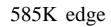


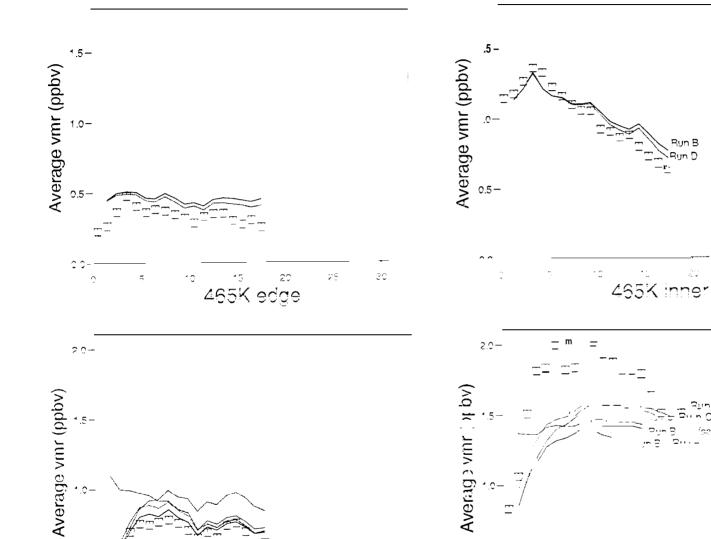
Figure 4



0.5

585K inner

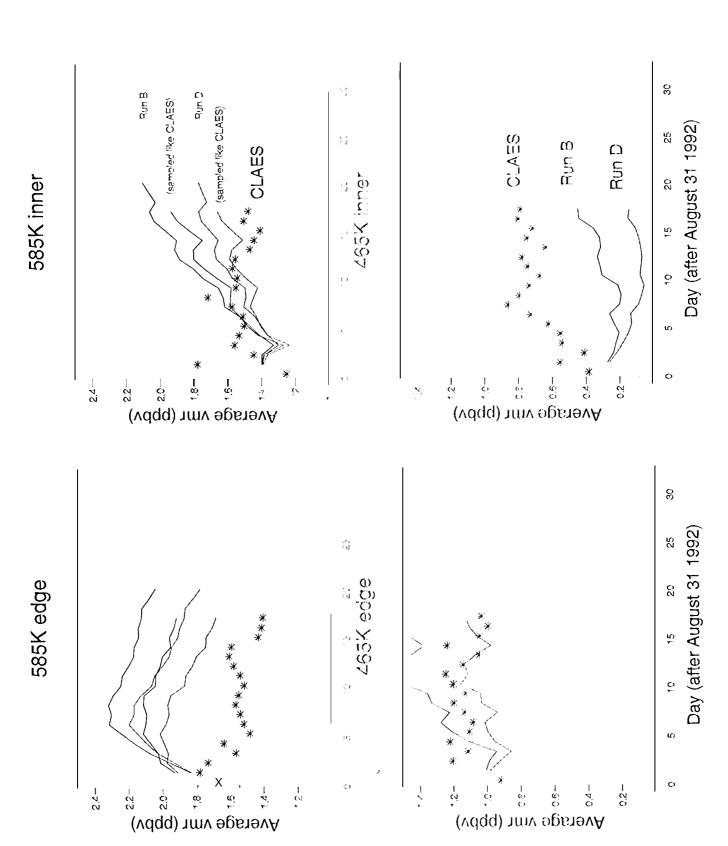
Day (after August 311992)



Day (after August 31 1992)



0.5-



o water

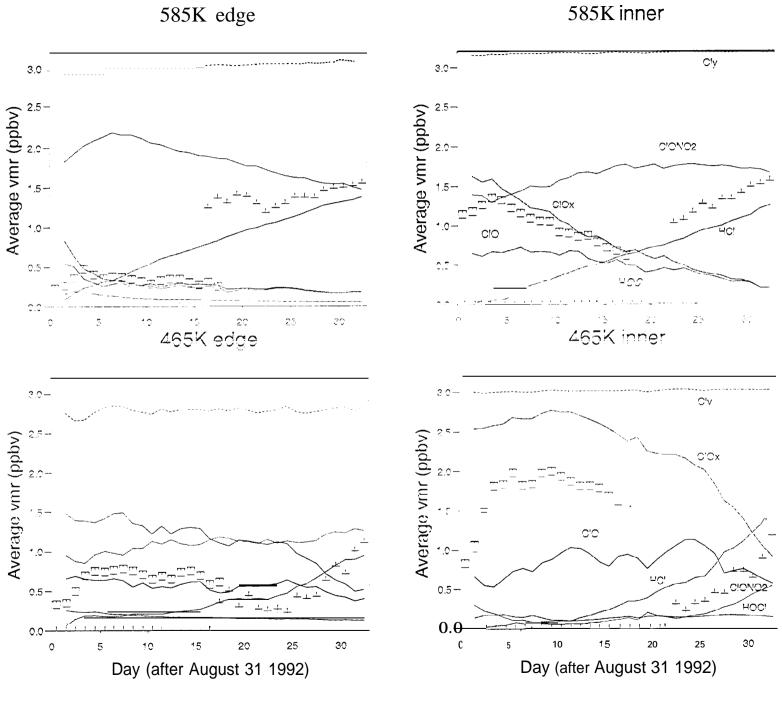


Figure 7

Lyere ga

HALOE data 585 K

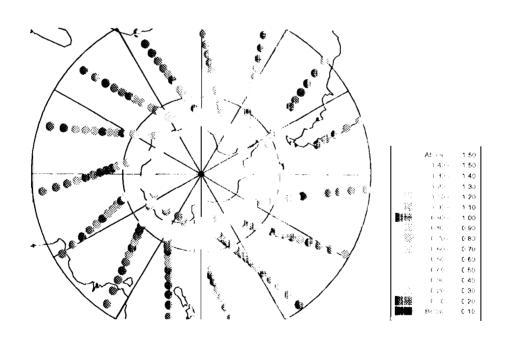
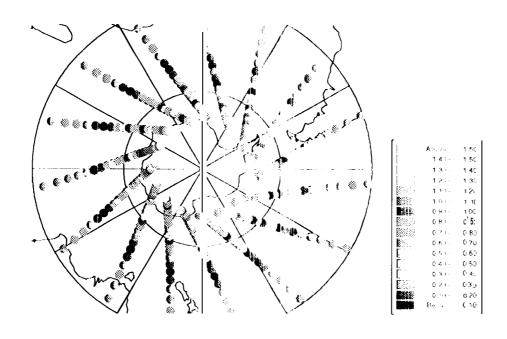


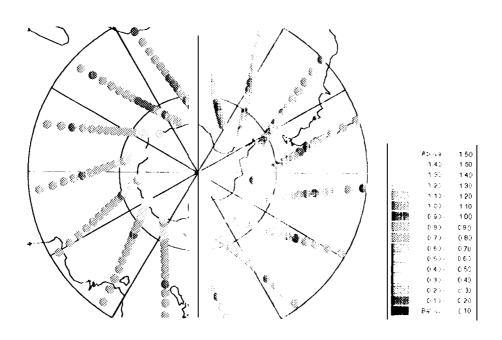
Figure 9)

HALOE data 465 K

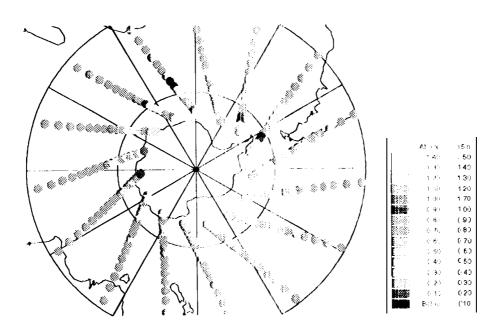


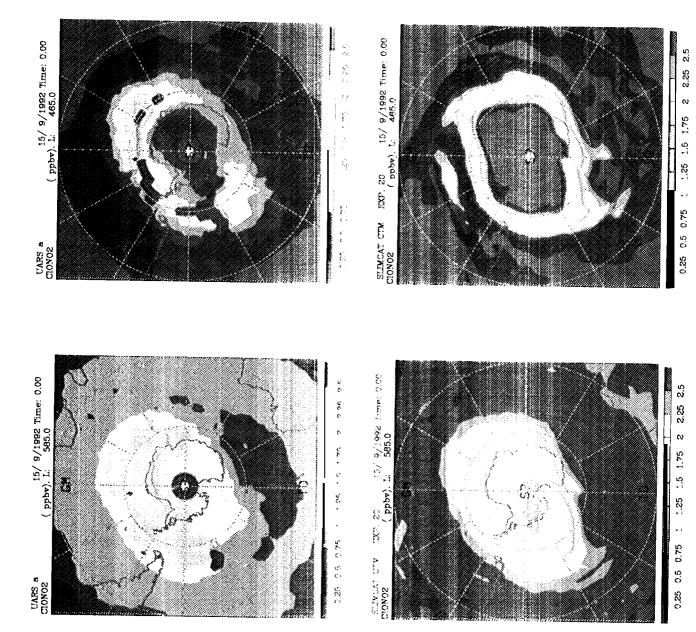
regime 90

CTM HCL expD 585 K



CTM HCL exp D 465 K





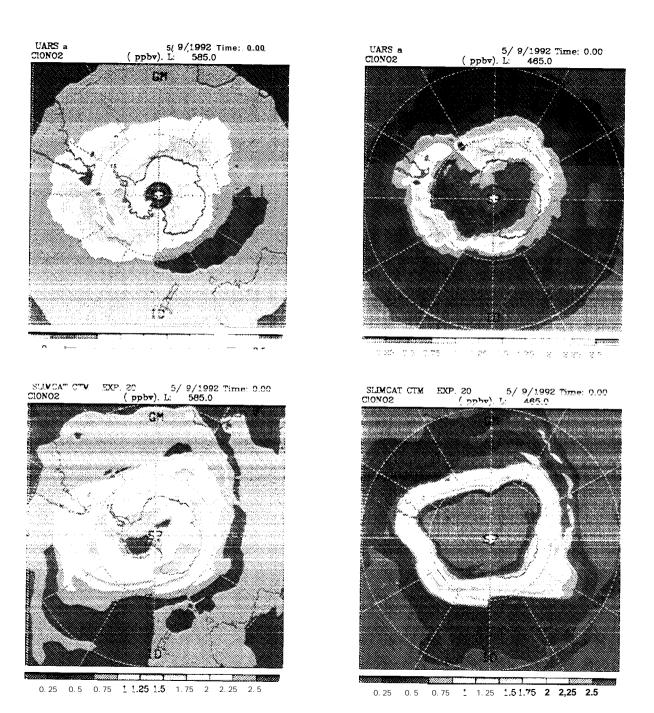
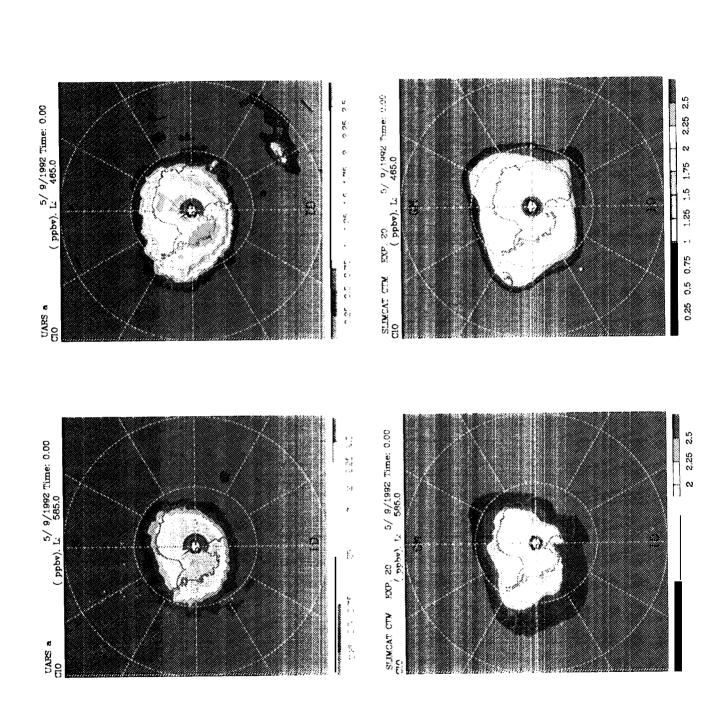


Figure to col



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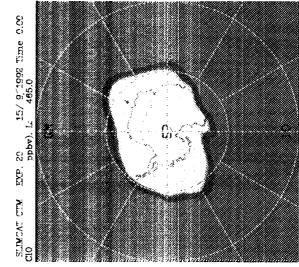
100

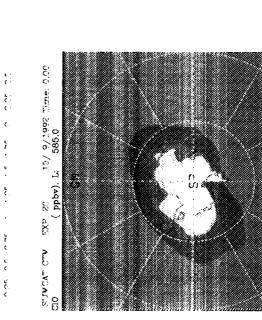
15/ 9/1992 Time: 0.00 ppbv). L: 465.0

UARS a

15/ 9/1992 Time: 0.00 (ppbv). L: 585.0

UARS a





0.25 0.5 0.75 1 1.25 1.5 1.75 2 2.25 2.5

0.25 0.5 0.75 1 1.25 1.5 1.75 2 2.25 2.5

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